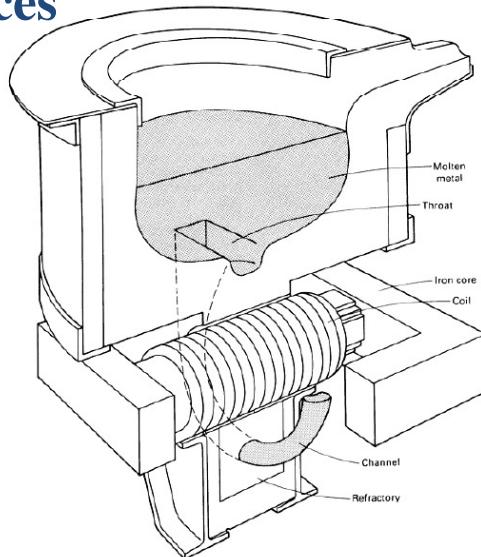
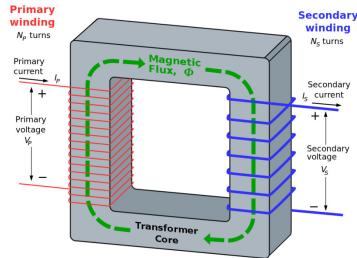


Channel Induction Furnace Induction Furnaces

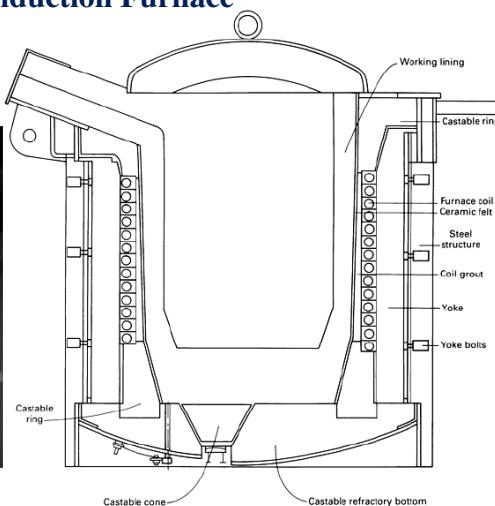
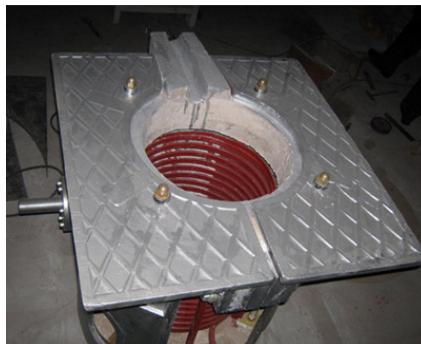
Transformer



Metals Handbook, Vol. 15
<https://wikipedia.org>

Fig. 2 A cross section of a channel-type induction furnace showing the water-cooled copper induction coil, which is located inside of a 360° loop formed by the throat and channel portion of the molten metal vessel. It is the channel portion of the loop, which serves as the secondary of the electrical circuit in which the copper coil is the primary.

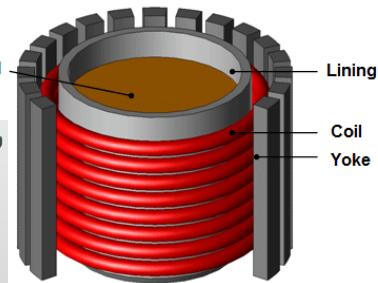
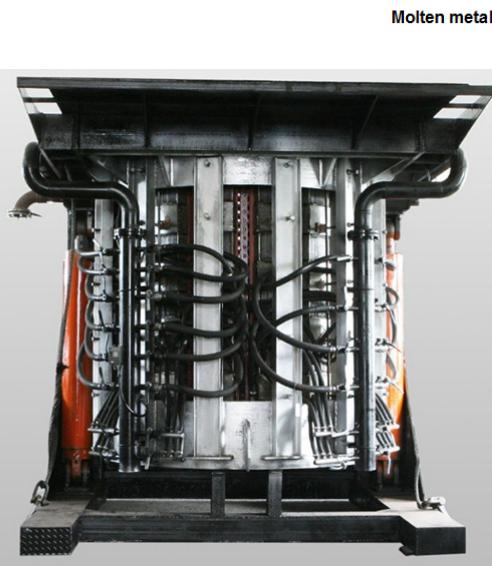
Coreless Induction Furnace



Metals Handbook, Vol. 15
www.psr-magnagroup.com

Fig. 1 A cross section of a coreless-type induction furnace showing water-cooled copper induction coil and key structural components. The entire molten metal bath (which serves as the secondary) is surrounded by the coil (the primary) that encircles the working lining.

Coreless Induction Furnace



<https://www.jmag-international.com>
<http://www.semigroup.com>

Coreless Induction Furnace



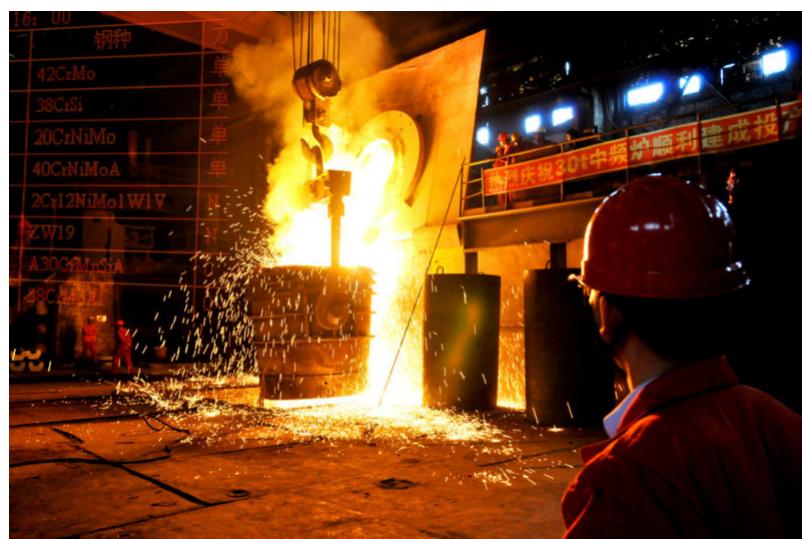
Atl-chatt.org

Coreless Induction Furnace



<http://www.tradeindia.com/>

Coreless Induction Furnace



<http://www.induction-heaters.net/>

Coreless Induction Furnace

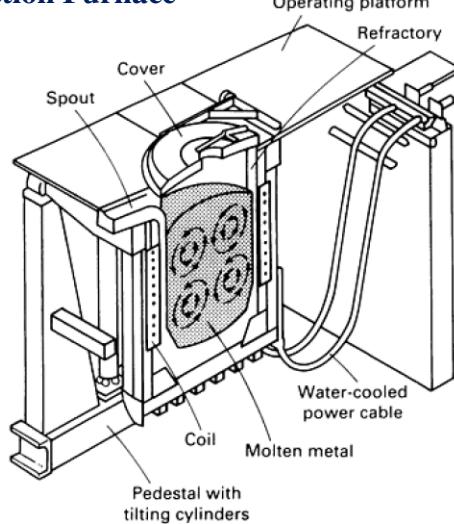


Fig. 3 A cross-sectional view of a coreless-type induction furnace illustrating four-quadrant stirring action, which aids in producing homogeneous melt

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Channel Induction Furnace

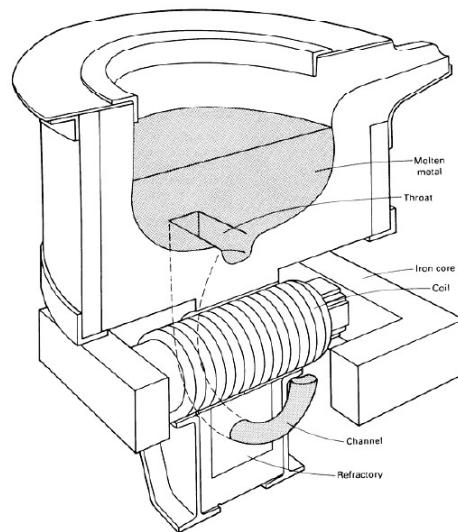
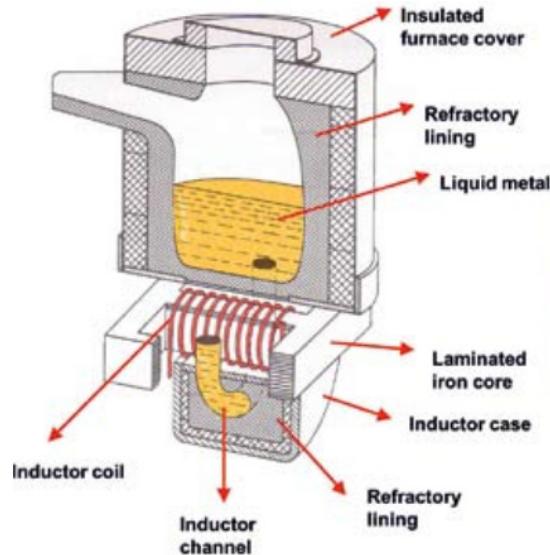


Fig. 2 A cross section of a channel-type induction furnace showing the water-cooled copper induction coil, which is located inside of a 360° loop formed by the throat and channel portion of the molten metal vessel. It is the channel portion of the loop, which serves as the secondary of the electrical circuit in which the copper coil is the primary.

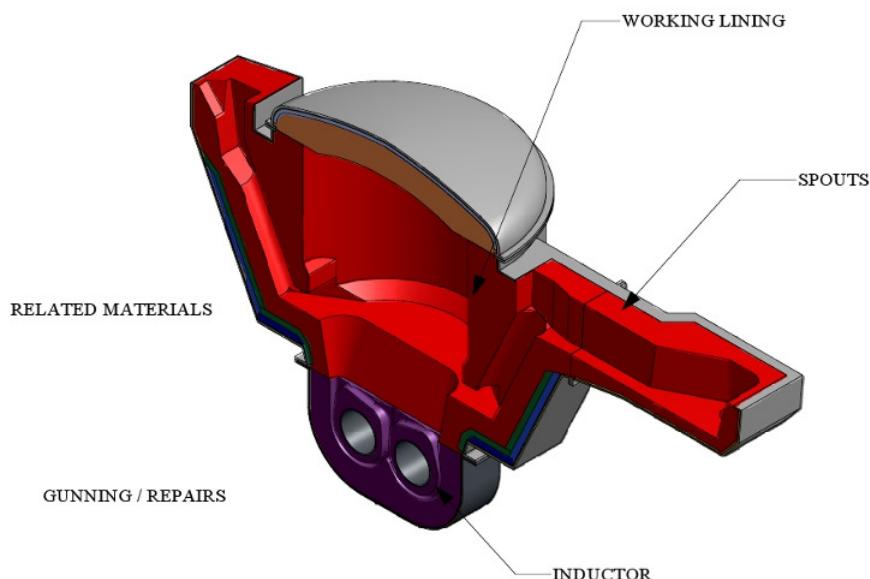
Metals Handbook, Vol. 15

Channel Induction Furnace



<http://www.marx-gmbh.de>

Channel Induction Furnace (Pressurized)



<http://www.alliedmineral.com>

Channel Induction Furnace



<http://www.erediscabini.com>

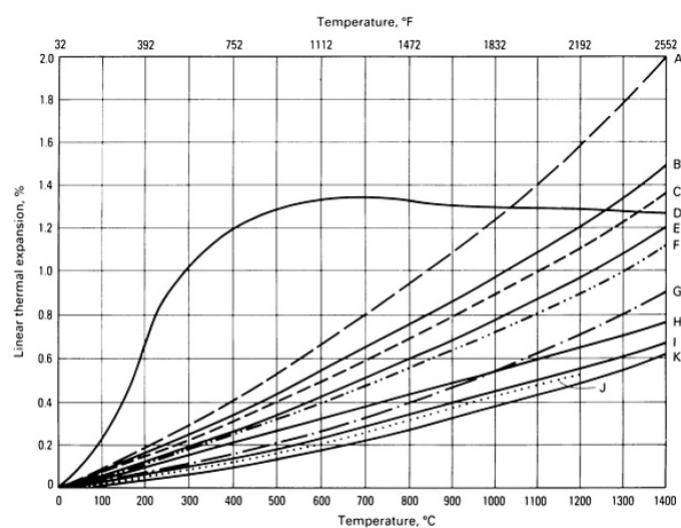
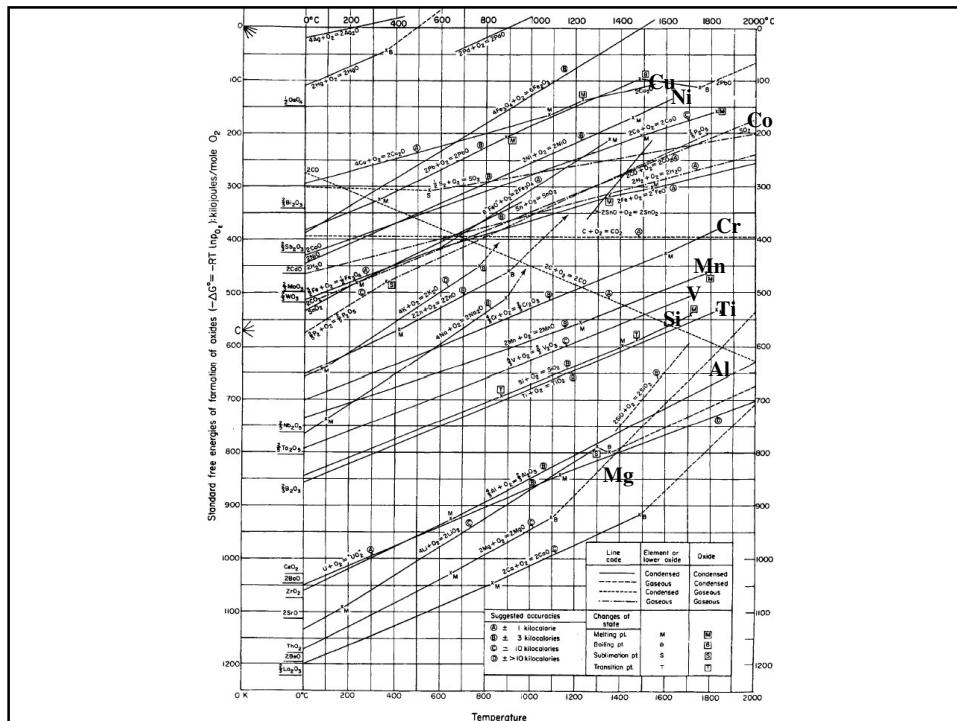


Fig. 7 Thermal expansion curves of various refractory brick oxide materials used for linings in induction furnaces: A, magnesia; B, chrome magnesia; C, chromite; D, silica; E, zirconia; F, corundum 99; G, corundum 90; H, fireclay; I, sillimanite; J, zircon; K, silicon carbide



Charge calculation

Table 49.
Carbon Content of Iron Versus Casting Wall Thickness

Wall thickness, mm	Carbon content, % in casting iron			Wall thickness, mm	Carbon content, % in casting iron			
	Si		into sand molds		Si		into sand molds	
	into permanent molds	with cores			without cores	into permanent molds		
6-10	2.2-2.6	3.0-3.3	2.8-3.0	21-40	1.6-2.0	2.6-2.8	2.2-2.4	
11-20	1.8-2.2	2.8-3.0	2.4-2.7	41-80	1.5-1.8	2.4-2.6	2.0-2.4	

The mass of gates varies in the range from 20 to 80% for small castings, 15 to 25% for medium-size castings, and 5 to 15% for large castings.

An example of charge calculation by a trial-and-error method. The gray iron used for casting automotive parts must have the following composition: 3.2-3.4% C, 2.0-2.2% Si, 0.6-0.8% Mn, about 0.15% P and 0.12% S. The melting loss in the cupola is 15% Si and 20% Mn; the pickup of sulfur comes to 50%. The chemical composition of charge materials is given in Table 50. The mass of metal charge is 800 kg.

Composition of Charge Components, %

Component	C	Si	Mn	P	S
Pig iron:					
JK1	3.5	3.3	0.50	0.11	0.02
JK2	3.6	3.0	0.50	0.12	0.03
Foundry returns	3.3	2.1	0.70	0.10	0.09
Steel scrap	0.2	0.3	0.80	0.05	0.05
Briquetted iron chips	3.3	2.1	0.70	0.10	0.09

Pig iron:

Content, kg

Total 800

Calculation of the Contents of Elements in the Charge

Calculation of the Contents of Elements in the Charge								
Component	Mass		Element content					
	kg	%	C	Si	Mn	P		S
Pig iron JK1	200	25	0.25×3.5 = 0.875	0.25×3.3 = 0.825	0.25×0.5 = 0.125	0.25×0.11 = 0.0275		$0.25 \times 0.02 = 0.005$
JK2	120	15	0.15×3.6 = 0.540	0.15×3.0 = 0.45	0.15×0.5 = 0.075	0.15×0.12 = 0.018		$0.15 \times 0.03 = 0.0045$
Foundry returns	240	30	0.30×3.3 = 0.990	0.3×2.1 = 0.630	0.3×0.7 = 0.210	0.30×0.10 = 0.03		$0.30 \times 0.09 = 0.027$
Briquetted chips	120	15	0.15×3.3 = 0.450	0.15×1.1 = 0.150	0.15×0.7 = 0.105	0.15×0.10 = 0.015		$0.15 \times 0.09 = 0.0135$
Steel scrap	120	15	0.15×0.20 = 0.03	0.15×0.30 = 0.045	0.15×0.8 = 0.120	0.15×0.05 = 0.0075		$0.15 \times 0.05 = 0.0075$
Total	800	100	2.93	2.01	0.635	0.098		$0.0575 + 0.0287 = 0.0862^{**}$

* Pickup of sulfur from coke is 50%.

- * Pickup of sulfur from coke is 50%.
- ** Sulfur content of iron to be melted will be below the specified value.

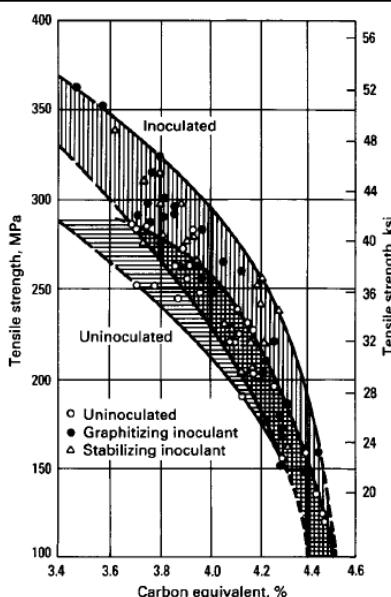


Fig. 6 Influence of inoculation on tensile strength of gray irons as a function of carbon equivalent for 30 mm (1.2 in.) diam bars. Source: Ref 7

ASM International, Alloying: Understanding the Basics, 2001

Table 2 Effects, levels, and sources of some trace elements in gray iron

Element	Trace level, %	Effects	Sources
Aluminum	≤0.03	Promotes hydrogen pinhole defects, especially when green sand molds are used and at levels above 0.005%. Neutralizes nitrogen	Deliberate addition, ferrous alloys, inoculants, scrap contaminated with aluminum components
Antimony	≤0.02	Promotes pearlite. Addition of 0.01% reduces the amount of ferrite sometimes found adjacent to cored surfaces	Vitreous enameled scrap, steel scrap, white metal bearing shells, deliberate addition
Arsenic	≤0.05	Promotes pearlite. Addition of 0.05% reduces the amount of ferrite sometimes found adjacent to cored surfaces	Pig iron, steel scrap
Bismuth	≤0.02	Promotes carbides and undesirable graphite forms that reduce tensile properties	Deliberate addition, bismuth-containing molds and core coatings
Boron	≤0.01	Promotes carbides, particularly in light-section parts. Effects become significant above about 0.001%	Deliberate addition, vitreous enameled scrap
Chromium	≤0.2	Promotes chill in thin sections	Alloy steel, chromium plate, some refined pig iron
Copper	≤0.3	Trace amounts have no significant effect and can be ignored	Copper wire, nonferrous alloys, steel scrap, some refined pig iron
Hydrogen	≤0.0004	Produces subsurface pinholes and (less often) fissures or gross blowing through a section. Mild chill promoter. Promotes inverse chill when insufficient manganese is present. Promotes coarse graphite	Damp refractories, mold materials, and additions
Lead	≤0.005	Results in Widmanstätten and "spiky" graphite, especially in heavy sections with high hydrogen. Can reduce tensile strength 50% at low levels (>0.0004%). Promotes pearlite	Some vitreous enamels, paints, free-cutting steels, nonferrous alloys, terne plate, white metal, solder, some pig irons
Molybdenum	≤0.05	Promotes pearlite	Some refined pig iron, steel scrap
Nickel	≤0.01	Trace amounts have no major effect and can be ignored	Refined pig iron, steel scrap
Nitrogen	≤0.02	Compacts graphite and increases strength. Promotes pearlite. Increases chill. Can cause pinhole and fissure defects. Can be neutralized by aluminum or titanium	Coke, carburizers, mold and core binders, some ferroalloys, steel scrap
Tellurium	≤0.003	Not usually found, but a potent carbide former	Free-cutting brasses, mold and core coatings, deliberate addition
Tin	≤0.15	Strong pearlite promoter; sometimes deliberately added to promote pearlitic structures	Solder, steel scrap, nonferrous alloys, refined pig iron, deliberate addition
Titanium	≤0.15	Promotes undercooled graphite. Promotes hydrogen pinholing when aluminum is present. Combines with nitrogen to neutralize its effects	Some pig irons, steel scrap, some vitreous enamels and paints, deliberate addition
Tungsten	≤0.05	Promotes pearlite	Tool steel
Vanadium	≤0.08	Forms carbides; promotes pearlite	Steel scrap; some pig irons

ASM International, Alloying: Understanding the Basics, 2001

Table 3 Effects of alloying elements on the mechanical and physical properties of gray iron

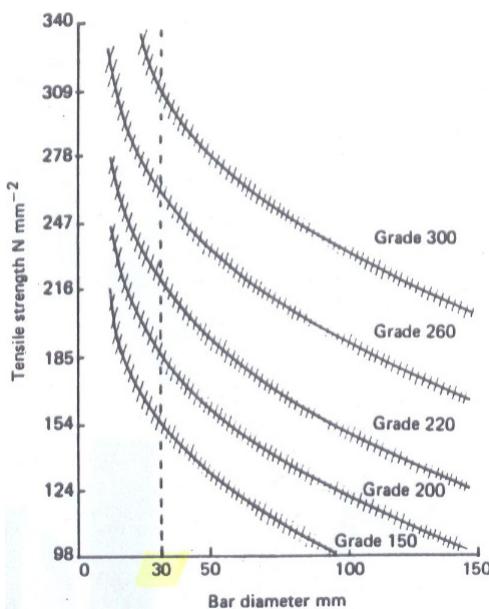
Alloying element	Effect of alloying element on:						
	Chill propensity	Pearlite stability	Machinability	Wear resistance	Hardness level	Hardenability	Strength
Silicon	Decreases	Decreases	Increases	Decreases	Decreases	Decreases	Decreases
Manganese	...	Increases	Increases	Increases	Increases
Chromium	Increases	Increases	Decreases	Increases	Increases	Increases	Increases
Molybdenum	Increases	Increases	Increases	Increases
Nickel	Decreases	Increases	Increases	Increases	Increases	Increases	Increases
Copper	...	Increases	Increases	Increases	Increases	...	Increases
Vanadium	Increases	Increases	Increases	Increases

Table 4 Compositions of ferrosilicon inoculants for gray cast iron

Performance category of inoculant	Composition, % ^(a)									
	Si	Al	Ca	Ba	Ce	TRE(b)	Ti	Mn	Sr	Others
Standard	46-50	0.5-1.25	0.60-0.90
	74-79	1.25 max	0.50-1.0
	74-79	0.75-1.5	1.0-1.5
	46-50	1.25 max	0.75-1.25	0.75-1.25	1.25 max
Intermediate	60-65	0.8-1.5	1.5-3.0	4-6	7-12
	70-74	0.8-1.5	0.8-1.5	0.7-1.3
	42-44	...	0.75-1.25	9-11
	50-55	...	5-7	9-11
High	50-55	...	0.5-1.5	9-11
	36-40	9-11	10.5-15
	46-50	0.50 max	0.10 max	0.60-1.0	...
	73-78	0.50 max	0.10 max	0.60-1.0	...
Stabilizing	6-11	0.50 max	0.50 max	48-52 Cr

(a) All compositions contain balance of iron. (b) TRE, total rare earth elements

ASM International, Alloying: Understanding the Basics, 2001

**Figure 1.6 Variation of tensile strength with section thickness for grey flake irons**

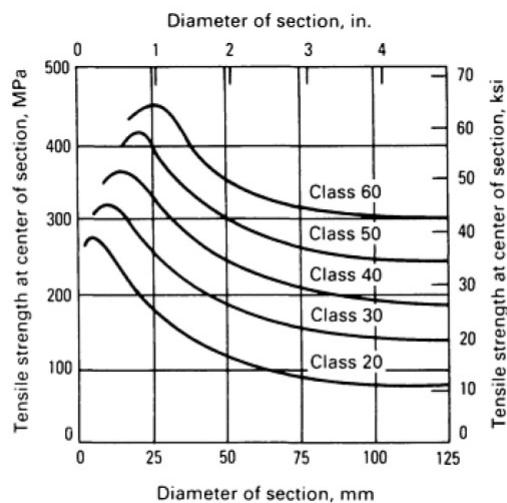


Fig. 12 Effect of section size on tensile strength of specimens cast from five classes of gray iron

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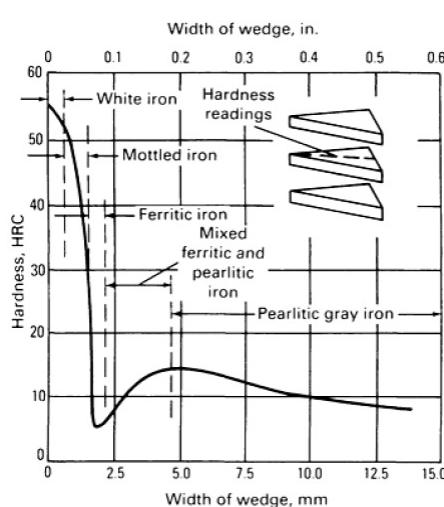


Fig. 11 Effect of section thickness on the hardness and microstructure of gray iron. Hardness readings were taken at increasing distance from the tip of a cast wedge (see inset). Iron composition was Fe-3.52C-2.55Si-1.01Mn-0.215P-0.086S.

International Comparison of Standard									مقایسه بین المللی در استاندارد	
Sweden Reference (SS)									ریفرانس ایکلور سوئد (SS)	
لیست معادلگیری در پارهای چدنی‌ای ریخته‌گری Material cross reference list										
Country:									کشور:	
Great Britain انگلستان	Sweden سوئد	USA آمریکا	Germany آلمان	France فرانسه	Italy ایتالیا	Spain اسپانیا	Japan ژاپن	استاندارد		
Standard:	BS	SS	AISI/SAE	W.-nr.	DIN	AFNOR	UNI	UNE	JIS	
چدن ریخته‌گری خاکستری / Gray cast iron /										
Grade 150	0100	1000	No 20B	0.6015	GG 15	F1 10 D			FC 100	
Grade 220	0110	No 25B	0.6015	GG 15	F1 15 D	G 15	FG 15	FC 150		
Grade 260	0120	No 30B	0.6020	GG 20	F1 20 D	G 20			FC 200	
Grade 300	0125	No 35B	0.6025	GG 25	F1 25 D	G 25	FG 25	FC 250		
Grade 350	0130	No 40B	0.6030	GG 30	F1 30 D	G 30	FG 30	FC 300		
Grade 400	0135	No 50B	0.6035	GG 35	F1 35 D	G 35	FG 35	FC 350		
L NiCuCr 202	0523	A 436 Type 2	0.6660	GG-L-NiCr 202	L-NC 202					
چدن ریخته‌گری باگرافیت کروی (ندولار) / Nodular cast iron /										
SNG 420/12	0717-02	60-40-18	0.7040	GGG 40	FCD 400-12	GS 370-17	FGE 38-17	FCD 400		
SNG 370/17	0717-12	—	—	GGG 40.3	FCD 370-17					
—	0717-15	—	0.7033	GGG 35.3	—					
SNG 500/7	0727-02	80-55-06	0.7050	GGG 50	FCD 500 - 7	GS 500	FGE 50-7	FCD 500		
Grade 56	0776	A43D2	0.7660	GGG-NiCr 202	S-NC 202	—	—	—		
SNG 600/3	0732-03	—	—	GGG 60	FCD 600-3	—				
SNG 700/2'	0737-01	100-70-03	0.7070	GGG 70	FCD 700-2	GS 700-2	FCD 700-2	FCD 700		

Type of iron	Total carbon, %	Silicon, %
Class 20 ...	3.40-3.60	2.30-2.50
Class 30 ...	3.10-3.30	2.10-2.30
Class 40 ...	2.95-3.15	1.70-2.00
Class 50 ...	2.70-3.00	1.70-2.00
Class 60 ...	2.50-2.85	1.90-2.10

Table 7 As-cast mechanical properties of standard gray iron test bars

Class	Tensile strength MPa	Tensile strength ksi	Torsional shear strength MPa	Torsional shear strength ksi	Compressive strength MPa	Compressive strength ksi	Reversed bending fatigue limit MPa	Reversed bending fatigue limit ksi	Transverse load on test bar kgf	Transverse load on test bar lbf	Hardness, HB
20	152	22	179	26	572	83	69	10	839	1850	156
25	179	26	220	32	669	97	79	11.5	987	2175	174
30	214	31	276	40	752	109	97	14	1145	2525	210
35	252	36.5	334	48.5	855	124	110	16	1293	2850	212
40	293	42.5	393	57	965	140	128	18.5	1440	3175	235
50	362	52.5	503	73	1130	164	148	21.5	1633	3600	262
60	431	62.5	610	88.5	1293	187.5	169	24.5	1678	3700	302

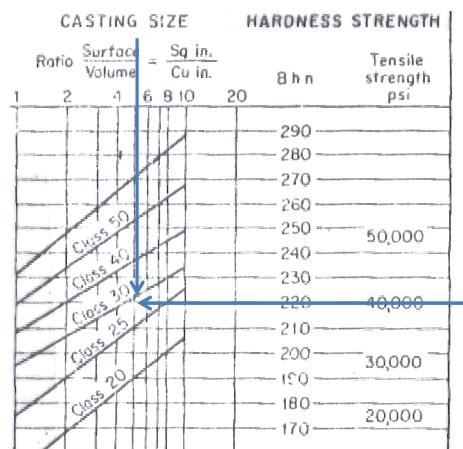


Fig. 20.8 Relationship of ASTM class number, tensile strength, hardness, and machinability of gray iron to ratio of casting surface area to volume.

Table 2 Hardness ranges for various combinations of gray iron microstructures

Microstructure	Hardness, HB
Ferrite + graphite.....	110–140
Pearlite + graphite.....	200–260
Pearlite + graphite + massive carbides...	300–450
Bainite + graphite.....	260–350
Tempered martensite + graphite.....	350–550
Austenite + graphite.....	140–160

Table 6 Typical pouring temperatures for some classes of gray iron

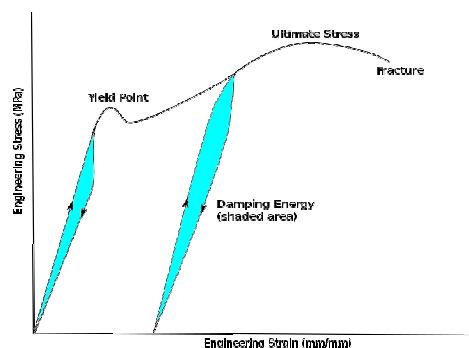
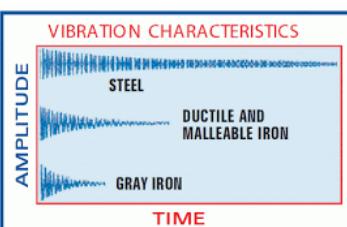
Class	Approximate liquidus temperature		Pouring temperature							
			Small castings				Large castings			
	°C	°F	°C	°F	°C	°F	°C	°F	°C	°F
30.....	1150	2100	1400	2550	1370	2500	1345	2450	1315	2400
35.....	1175	2150	1425	2600	1400	2550	1370	2500	1345	2450
40.....	1200	2190	1450	2640	1420	2590	1395	2540	1365	2490
45.....	1220	2230	1470	2680	1445	2630	1415	2580	1390	2530

Table 3 Minimum recommended section sizes for unalloyed gray irons

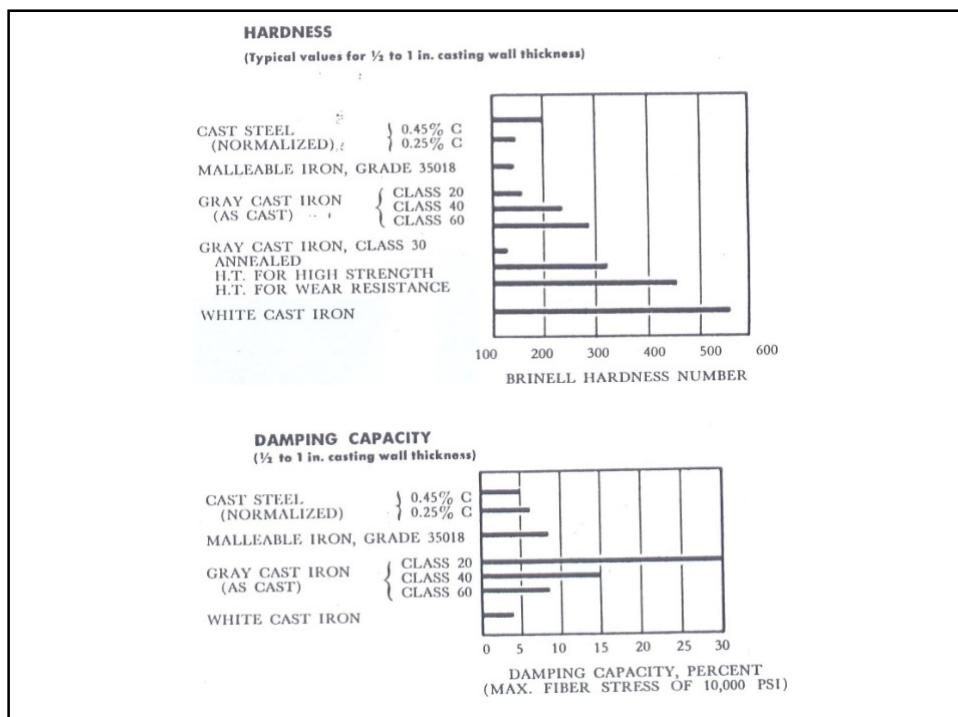
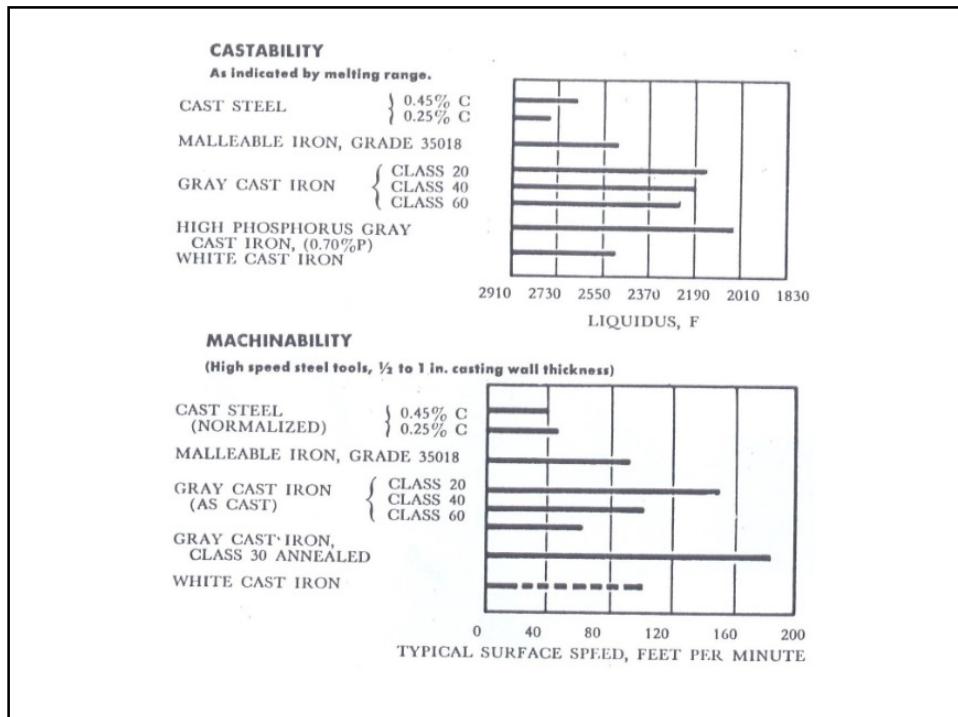
Class	Minimum section thickness mm	Minimum section thickness in.	Volume-to-surface-area ratio(a) mm	Volume-to-surface-area ratio(a) in.
20	3.2	1/8	1.5	0.06
25	6.4	1/4	3.0	0.12
30	9.5	3/8	4.3	0.17
35	9.5	3/8	4.3	0.17
40	15.9	5/8	7.1	0.28
50	19.0	1/2	8.4	0.33
60	25.4	1	10.7	0.42

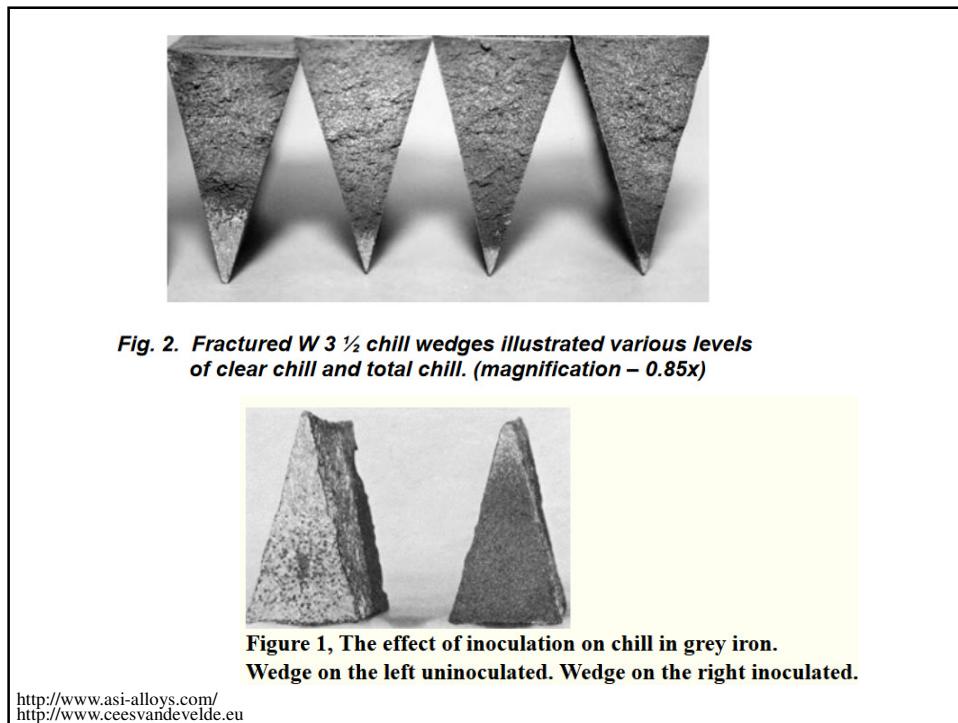
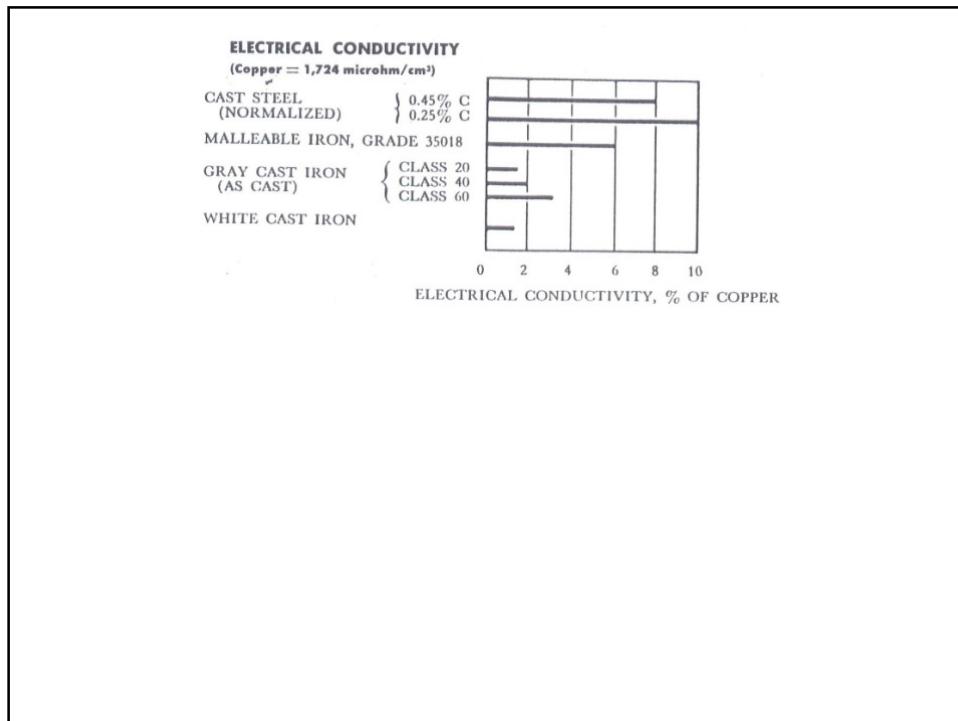
(a) Volume-to-surface-area ratios are for square plates.

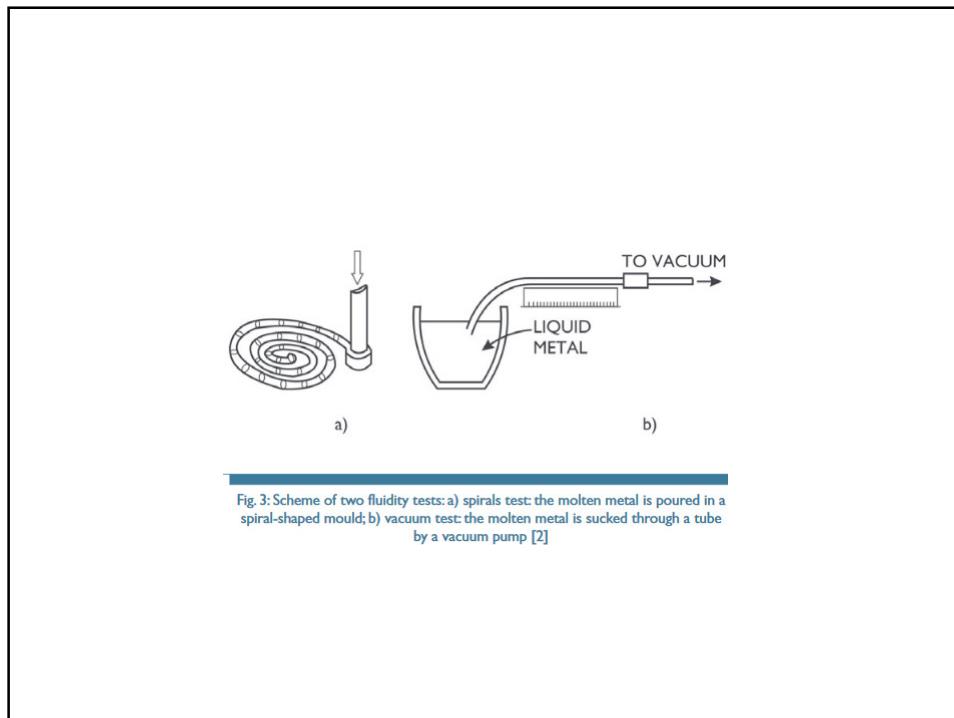
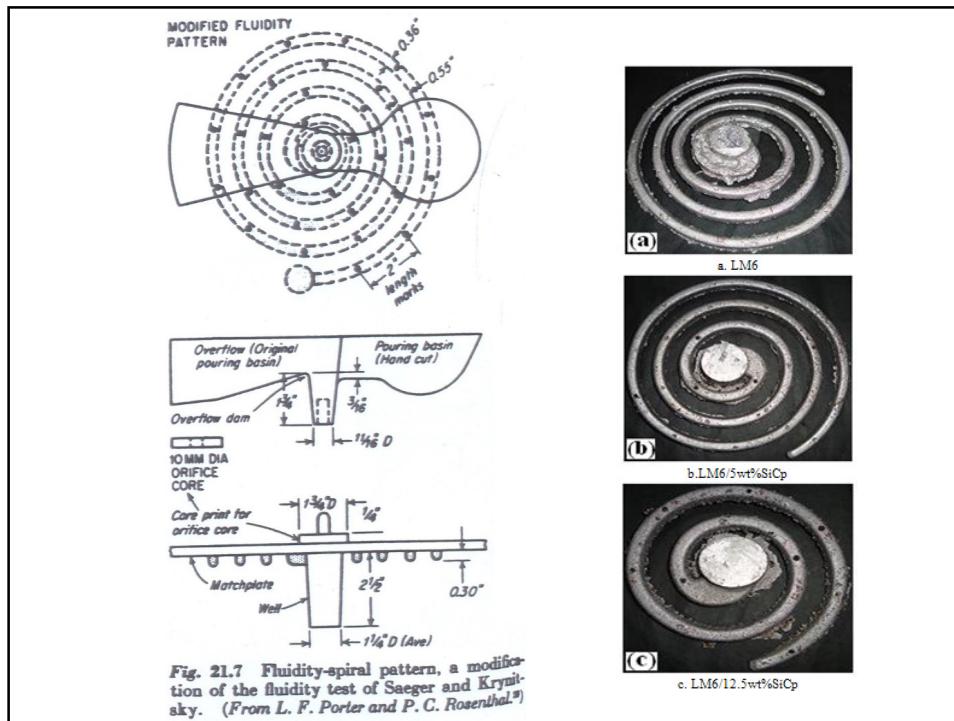
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Material	Relative damping capacity
White iron	2-4
Malleable iron	8-15
Ductile iron	5-20
Gray iron, fine flake	20-100
Gray iron, coarse flake	100-500
Eutectoid steel	4
Armco iron	5
Aluminum	0.4



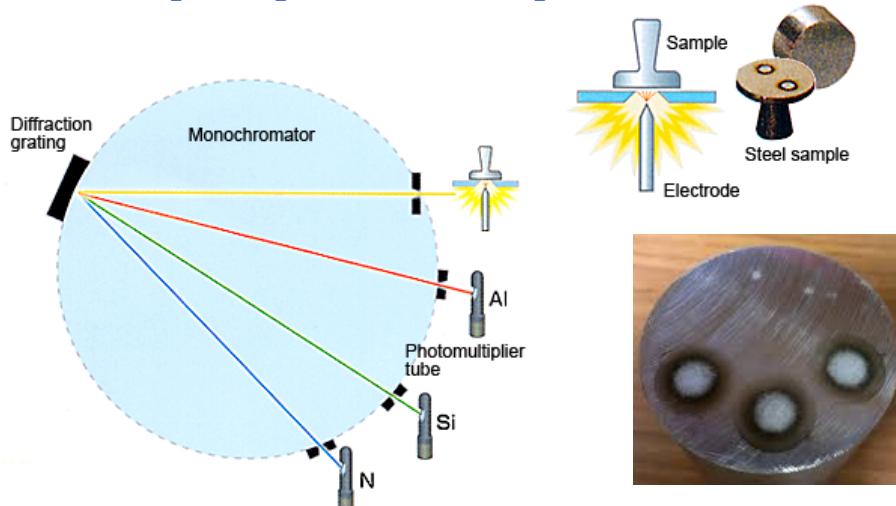




Spark optical emission spectrometer

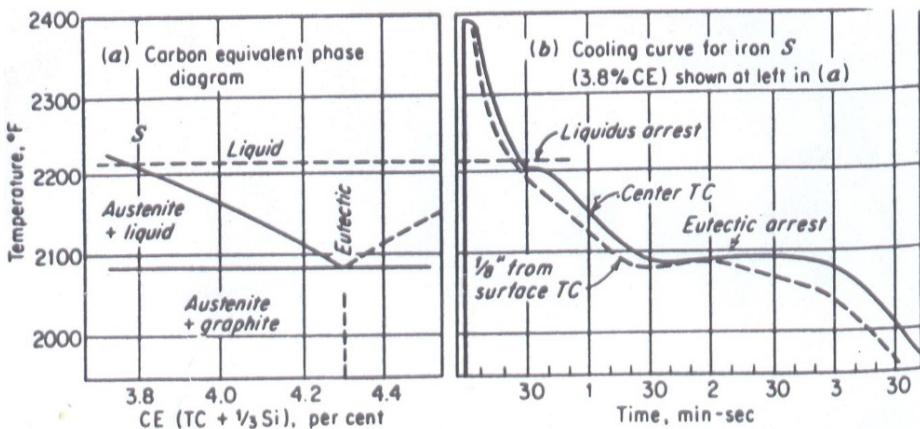


Spark optical emission spectrometer



<http://www.shimadzu.com>

Thermal analysis



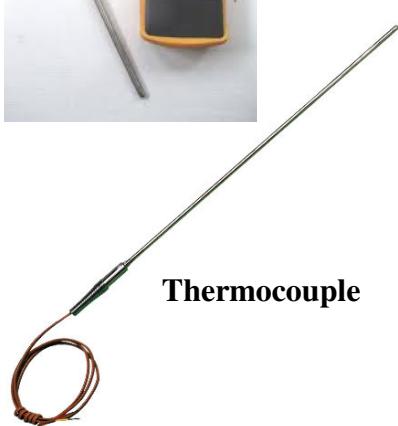
The relationship of the carbon-equivalent phase diagram in (a) to the thermal arrests on the cooling curve in (b).

Thermometer

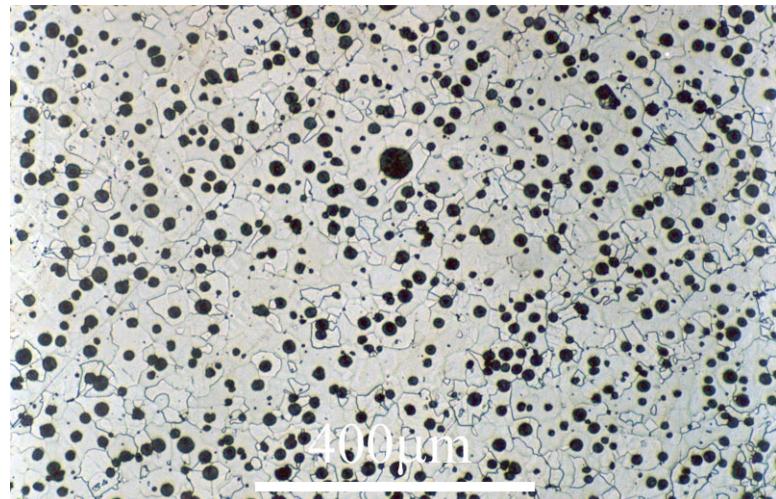


Pyrometer

Thermocouple



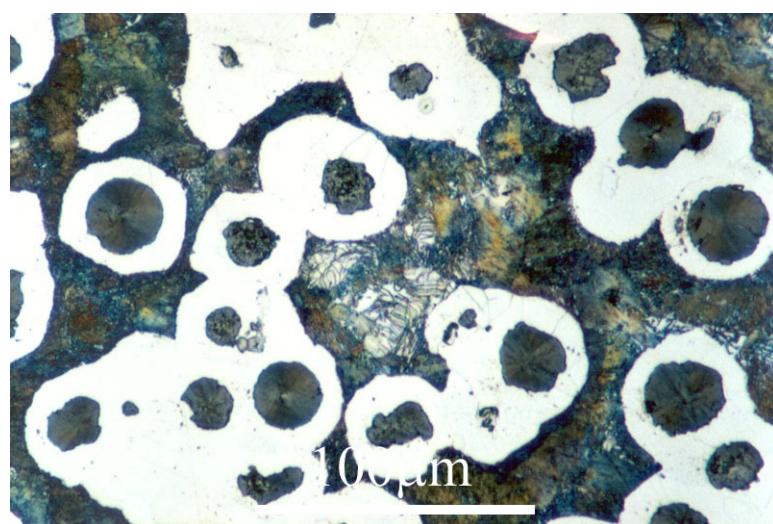
Ductile cast Iron



<http://www.doitpoms.ac.uk/miclib/index.php>

60

Ductile cast Iron



<http://www.doitpoms.ac.uk/miclib/index.php>

62

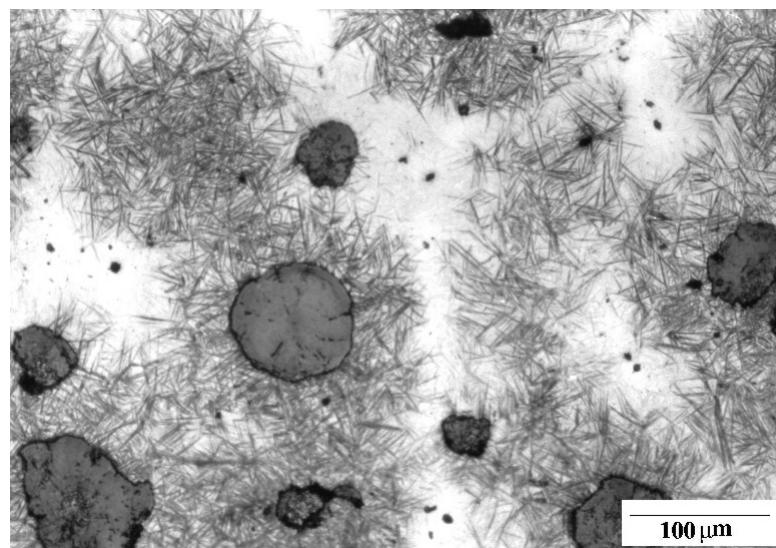
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<http://www.doitpoms.ac.uk/miclib/index.php>

63

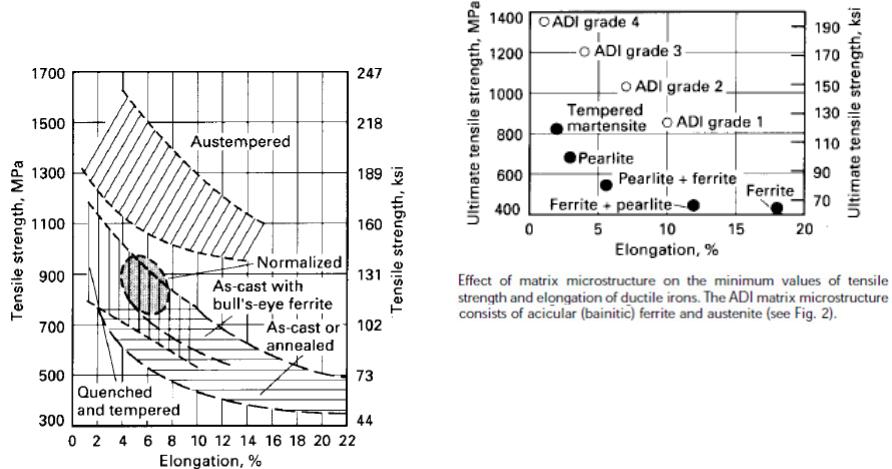
Austempered Ductile Iron (ADI)



<http://www.doitpoms.ac.uk/miclib/index.php>

65

Ductile cast Iron



Range of tensile strength and elongation values for as-cast and heat-treated ductile irons.

ASM International, Alloying: Understanding the Basics, 2001

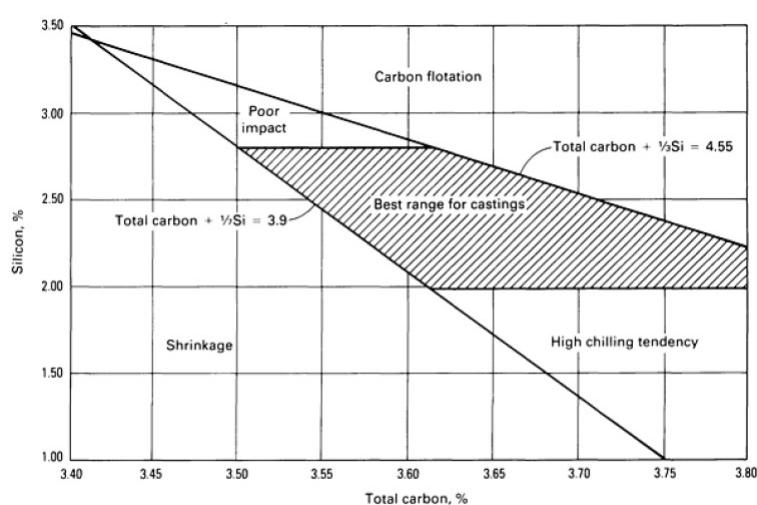


Fig. 1 Typical carbon and silicon ranges for ductile iron castings. Source: Ref 1

Table 2 Mechanical properties and typical applications for standard grades of ductile iron

Specification No.	Grade or class	Hardness, HB(a)	Tensile strength, min(b)		Yield strength, min(b)		Elongation in 50 mm (2 in.) (min, % (b))	Typical applications	
			MPa	ksi	MPa	ksi			
ASTM A395; ASME SA395	60-40-18	143-187	414	60	276	40	18	Valves and fittings for steam and chemical plant equipment	
ASTM A476(c); SAE AMS 5316	80-60-03	201 min	552	80	414	60	3	Paper mill dryer rolls	
ASTM A536	60-40-18	...	414	60	276	40	18	Pressure-containing parts, such as valve and pump bodies	
		65-45-12	...	448	65	310	45	12	Machine components subject to shock and fatigue loads
		80-55-06	...	552	80	379	55	6	Crankshafts, gears, and rollers
		100-70-03	...	689	100	483	70	3	High-strength gears and machine components
SAE J434		120-90-02	...	827	120	621	90	2	Pinions, gears, rollers, and slides
	D4018	170 max	414	60	276	40	18	Steering knuckles	
	D4512	156-217	448	65	310	45	12	Disk brake calipers	
	D5506	187-255	552	80	379	55	6	Crankshafts	
	D7003	241-302	689	100	483	70	3	Gears	
	DQ&T	(c)	(d)	(d)	(d)	(d)	(d)	Rocker arms	
SAE AMS 5315C	Class A	190 max	414	60	310	45	15	Electric equipment, engine blocks, pumps, housings, gears, valve bodies, clamps, and cylinders	

Note: For compositions, descriptions, and uses, see Table 1. (a) Measured at a predetermined location on the casting. (b) Determined using a standard specimen taken from a separately cast test block, as set forth in the applicable specification. (c) Range specified by mutual agreement between producer and purchaser. (d) Value must be compatible with minimum hardness specified for production castings.

Table 3 Hardness, toughness, and tensile properties at room temperature for austempered ductile iron grades specified in ASTM A 897 and A 897M (metric)

Grade	Minimum tensile strength		Minimum yield strength		Minimum elongation, %	Unnotched Charpy impact energy		Typical hardness, HB
	MPa	ksi	MPa	ksi		J	ft · lbf	
125/80/10 (Grade 1)	850	125	550	80	10	100	75	269-321
150/100/7 (Grade 2)	1050	150	700	100	7	80	60	302-363
175/125/4 (Grade 3)	1200	175	850	125	4	60	45	341-444
200/155/1 (Grade 4)	1400	200	1100	155	1	35	25	388-477
230/185/- (Grade 5)	1600	230	1300	185	444-555

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Table 4 Summary of the effects of various elements found in ductile irons

Element	Typical amount	Maximum for matrix		Positive effects	Deleterious effects
		Ferrite	Pearlite		
Spheroidizing elements					
Mg	0.02-0.08%	Sufficient to ensure spheroidal graphite	Lowers sulfur and oxygen contents; causes graphite to form spheroids		Excess promotes carbides
Rare earths	0-0.30%	About 0.035%	About 0.035%	Promotes nodule count and quality in combination with magnesium; neutralizes subversive elements	Excess promotes carbides in thin sections and chunky graphite in heavy sections
Ca	Not detected	Essentially insoluble	Essentially insoluble	Increases nodule count and improves nodule quality; optimizes inoculation	Excess promotes carbides
Ba	Not detected	Essentially insoluble	Essentially insoluble	Increases nodule count; optimizes inoculation	...
Primary elements					
C	3.00-4.00%	3.00-4.00%	3.00-4.00%	Present as spheroids or carbides	Excess results in graphite formation
Si	1.80-3.00%	1.80-3.00%	1.80-2.75%	Promotes graphitization during solidification and matrix formation	Hardens and strengthens ferrite; increases ni-ductility
P	About 0.02%	0.035% max	0.05% max	Kept as low as possible	Forms intercellular carbide network; promotes pearlite
S	0.01-0.02%	0.02% max	0.02% max	Combines with magnesium and rare earths	Limits efficiency of magnesium treatment process
Mn	0.00-1.20%	0.20%	0.80% max	Promotes pearlite in as-cast and normalized iron	Intercellular carbides when over 0.70%
Alloying elements					
Ni	0.01-2.00%	As low as possible for as-cast	To specification	Employed for hardenability (e.g., pearlitic)	...
Mo	0.01-0.75%	0.03% max	To specification	Promotes hardenability	Excess promotes intercellular carbides
Cu	0.01-0.90%	0.03% max	To specification	Promotes pearlitic hardenability	No significant effect on nodule count or quality
Tramp elements					
Te	<0.005%	0.02% max	0.02% max	Used to control pinholes	Promotes spheroid degeneration in absence of rare earths
Pb	...	0.002% max	0.002% max	Kept as low as possible	Promotes intercellular flake graphite
Ti	<0.07%	0.03% max	0.07% max	Kept as low as possible	Promotes vermicular graphite
Al	0.003-0.06%	0.05% max	0.05% max	Used in ferroalloys to suppress chill	Promotes vermicular graphite effect—greater in heavy sections; promotes ninnites

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Sb	<0.005%	0.001% max	0.001% max	Strong pearlite former; counteracts chunky graphite in heavy sections	Nodule degeneration at high levels when rare earths are not present
Bi	<0.01%	0.002% max	0.002% max	Increases nodule count and quality when rare earths are present	Promotes vermicular graphite in absence of rare earths
Zr	<0.01%	0.10% max	0.10% max	Kept as low as possible	Promotes vermicular graphite
Carbide- and pearlite-forming elements					
Cr	0.02–0.15%	0.04% max	0.10% max	Very powerful carbide former	Carbides resistant to annealing
B	<0.0005%	0.002% max	0.002% max	Kept as low as possible	Forms intercellular carbides that resist annealing
Sn	<0.10%	0.01% max	0.08% max	Very potent pearlite former	At <0.10%, forms intercellular structure with flake graphite
As	0.01% max	0.02% max	0.05% max	About 0.08% required for pearlitic matrix	...
V	<0.04%	0.04% max	0.05% max	Forms very stable carbides	Retards annealing
Gaseous elements					
O	<0.005%	About 0.003%	About 0.003%	Kept as low as possible	Combines with magnesium
H	0.0002–0.0015%	About 0.0003%	About 0.0003%	Kept as low as possible	Promotes centerline carbides and inverse chill; promotes pinholes
N	Kept as low as possible	Mild carbide-forming tendency; may contribute to porosity

Source: Ref 1

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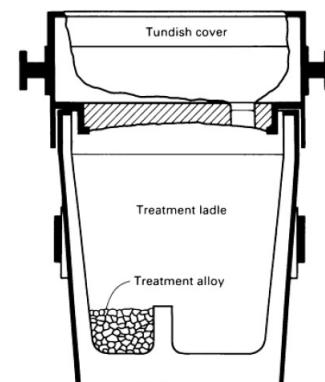
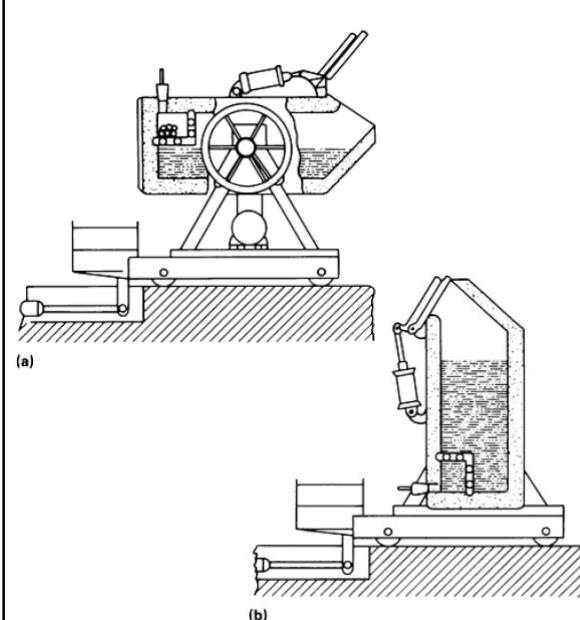


Fig. 3 Treatment ladle with tundish cover used in the magnesium treatment of ductile iron

Fig. 2 Schematic of the Fischer converter. (a) Vessel in filling position. (b) Vessel in treating position
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Compacted Graphite cast Iron

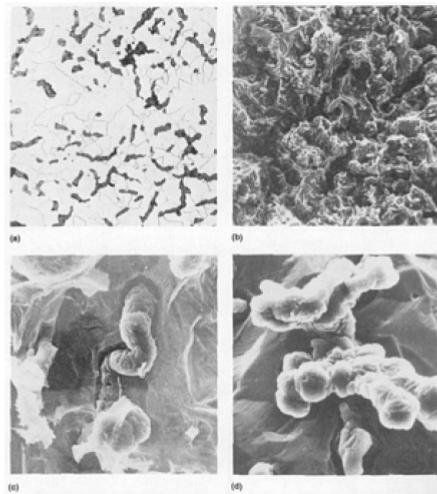


Fig. 1 Typical microstructures of CG irons. (a) Optical micrograph. Etched with nital. (b) Tensile load fracture surface. Overall view. Ion bombardment etched. SEM, 65 \times . (c) and (d) Examples of true shape of graphite in CG irons. Full deep etch. SEM, 395 \times . Courtesy of Austrian Foundry Research Institute

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Compacted Graphite cast Iron

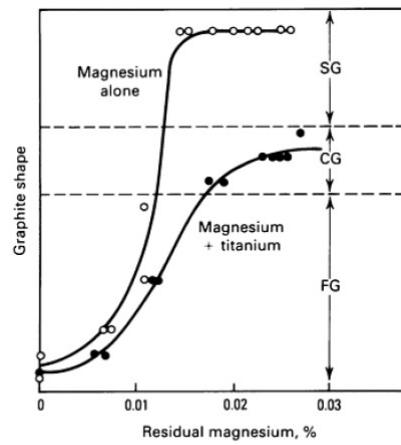


Fig. 6 Range of residual magnesium that produces compacted graphite. Source: Ref 5

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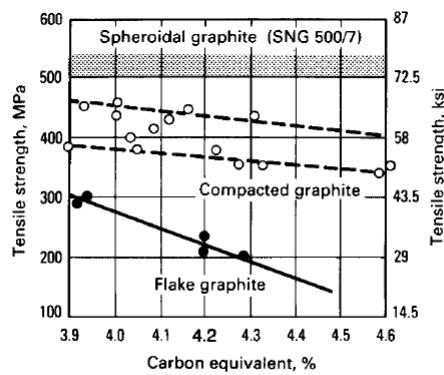


Fig. 6 Effect of carbon equivalent on the tensile strength of flake, compacted, and spheroidal graphite irons cast in 30 mm (1.2 in.) diam bars

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